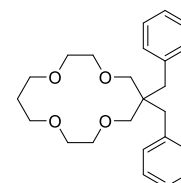


## Product Information



### 62567 Lithium ionophore VI

(6,6-Dibenzyl-14-crown-4; 6,6-Dibenzyl-1,4,8-11-tetraoxacyclotetradecane)

Selectophore®, function tested

## Electrochemical Transduction

### Ion-Selective Electrodes

#### Application 1 and Sensor Type<sup>1,2,3</sup>

Neutral carrier for Li<sup>+</sup> in ion-selective PVC membrane electrodes with high selectivity over potassium and sodium ions.

#### Recommended Membrane Composition

1.00 wt%	Lithium ionophore VI ( <a href="#">62567</a> )
0.70 wt%	Potassium tetrakis(4-chlorophenyl)borate ( <a href="#">60591</a> )
70.30 wt%	2-Nitrophenyl octyl ether ( <a href="#">73732</a> )
28.00 wt%	Poly(vinyl chloride) high molecular weight ( <a href="#">81392</a> )

#### Recommended Cell Assembly

Reference || sample solution || liquid membrane | 1 M LiCl | AgCl, Ag

#### Electrode Characteristics and Function

Selectivity coefficients  $\log K_{Li,M}^{Pot}$  determined by the fixed interference method (0.05 M solution for alkali metal ions and H<sup>+</sup>, 0.5 M for alkaline earth metal ions and NH<sub>4</sub><sup>+</sup>).

$\log K_{Li,H}^{Pot}$	-3.0	$\log K_{Li,NH_4}^{Pot}$	-3.0
$\log K_{Li,Na}^{Pot}$	-2.4	$\log K_{Li,Mg}^{Pot}$	-4.3
$\log K_{Li,K}^{Pot}$	-2.3	$\log K_{Li,Ca}^{Pot}$	-4.7

Slope of linear regression:

59-60 mV/dec

<sup>1</sup> Synthesis and Selectivity for Lithium of Lipophilic 14-Crown-4 Derivatives Bearing Bulky Substituents or an Additional Binding Site in the Side Arm K. Kimura, H. Yano, S. Kitazawa, T. Shono, J. Chem. Soc. Perkin Trans. II, 1945 (1986).

<sup>2</sup> Lithium Ion Selective Electrodes Based on Crown Ethers for Serum Lithium Assay. K. Kimura, O. Oishi, T. Miura, T. Shono, Anal. Chem. 59, 2331 (1987).

<sup>3</sup> Voltammetric Lithium Ion-Selective Electrodes Based on Ion Transfer at the Oil/Water Interface Facilitated by Neutral Ionophores. S. Sawada, T. Osakai, M. Senda, Anal. Sci. 11, 733 (1995).

